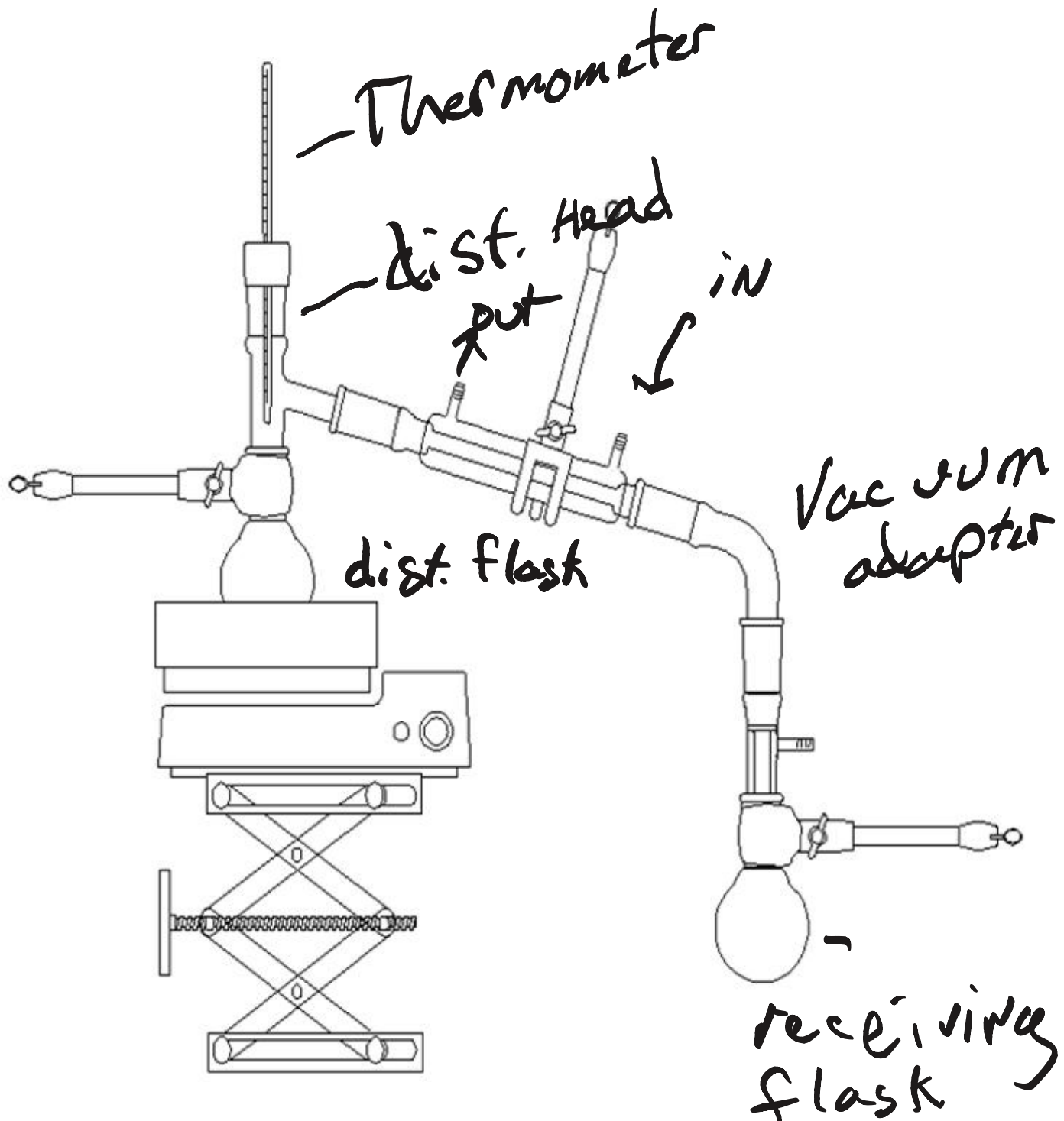


# Distillation

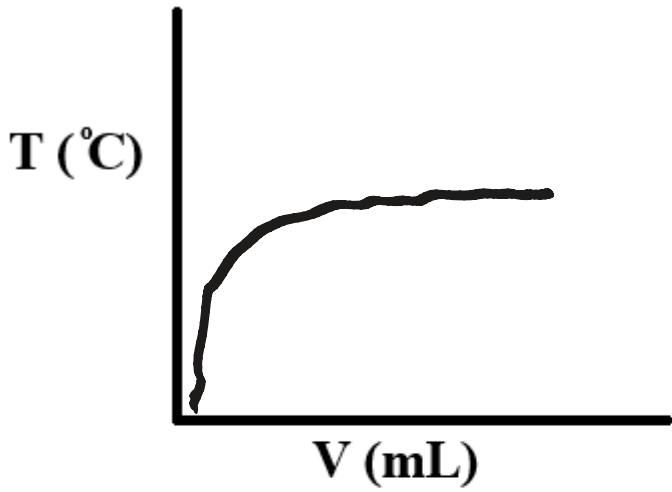
220C

6/8/26

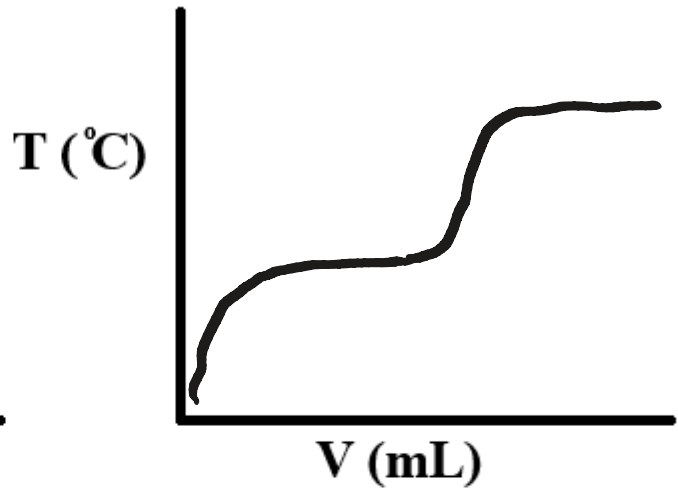
# Simple Distillation L/S



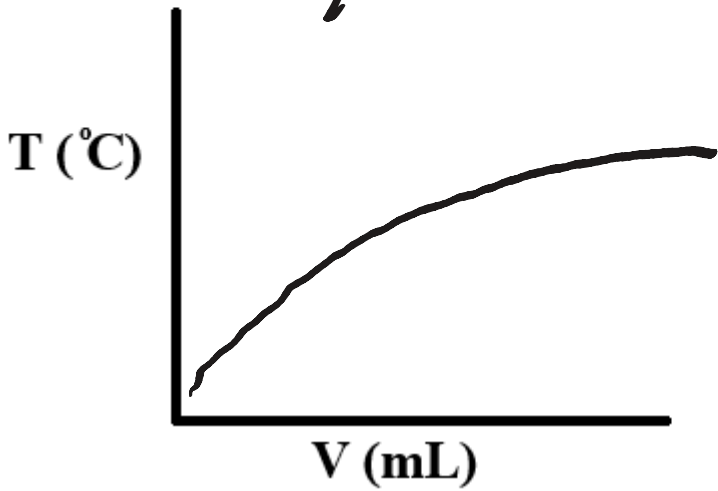
L/S



L/L 100°C

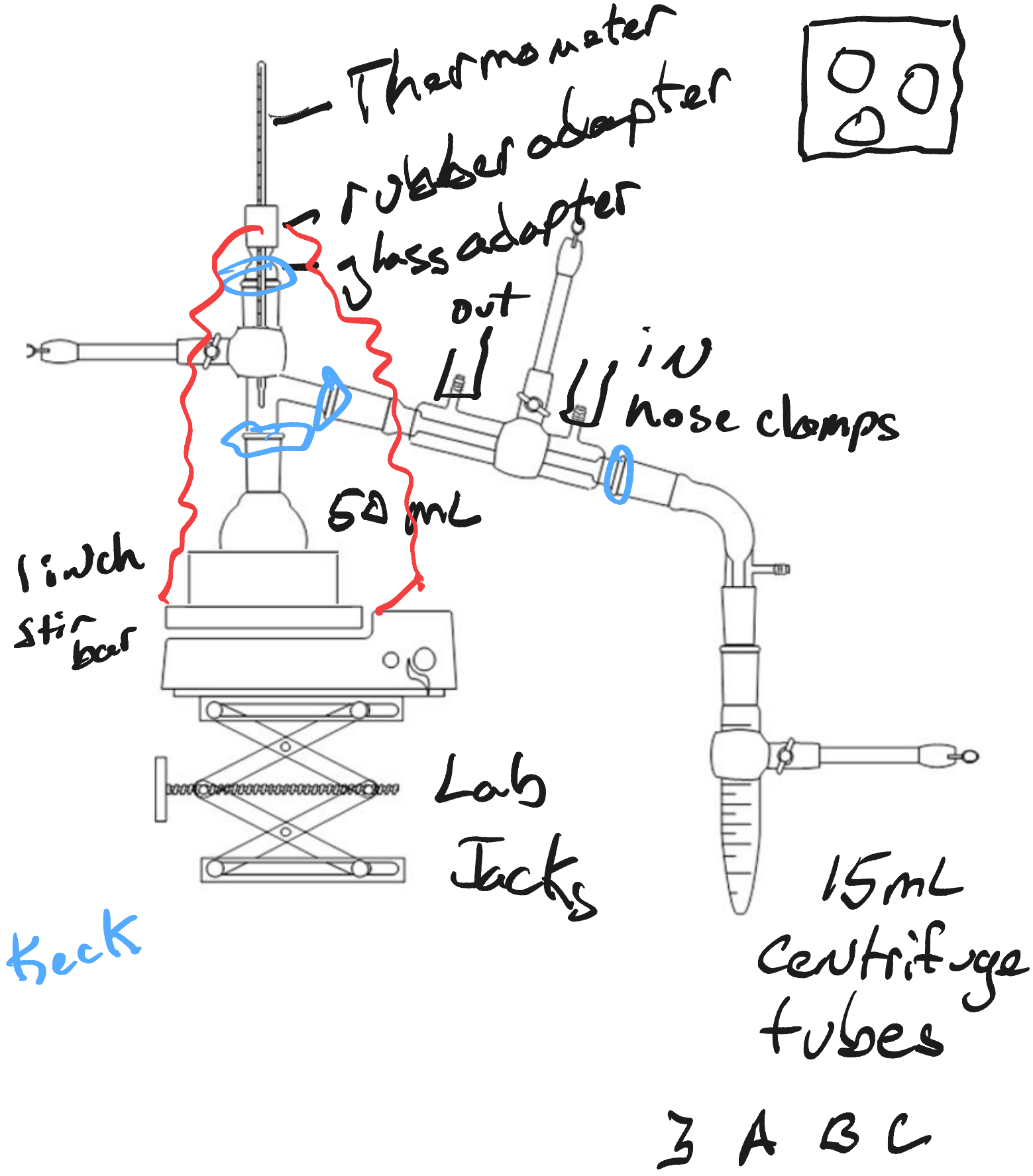


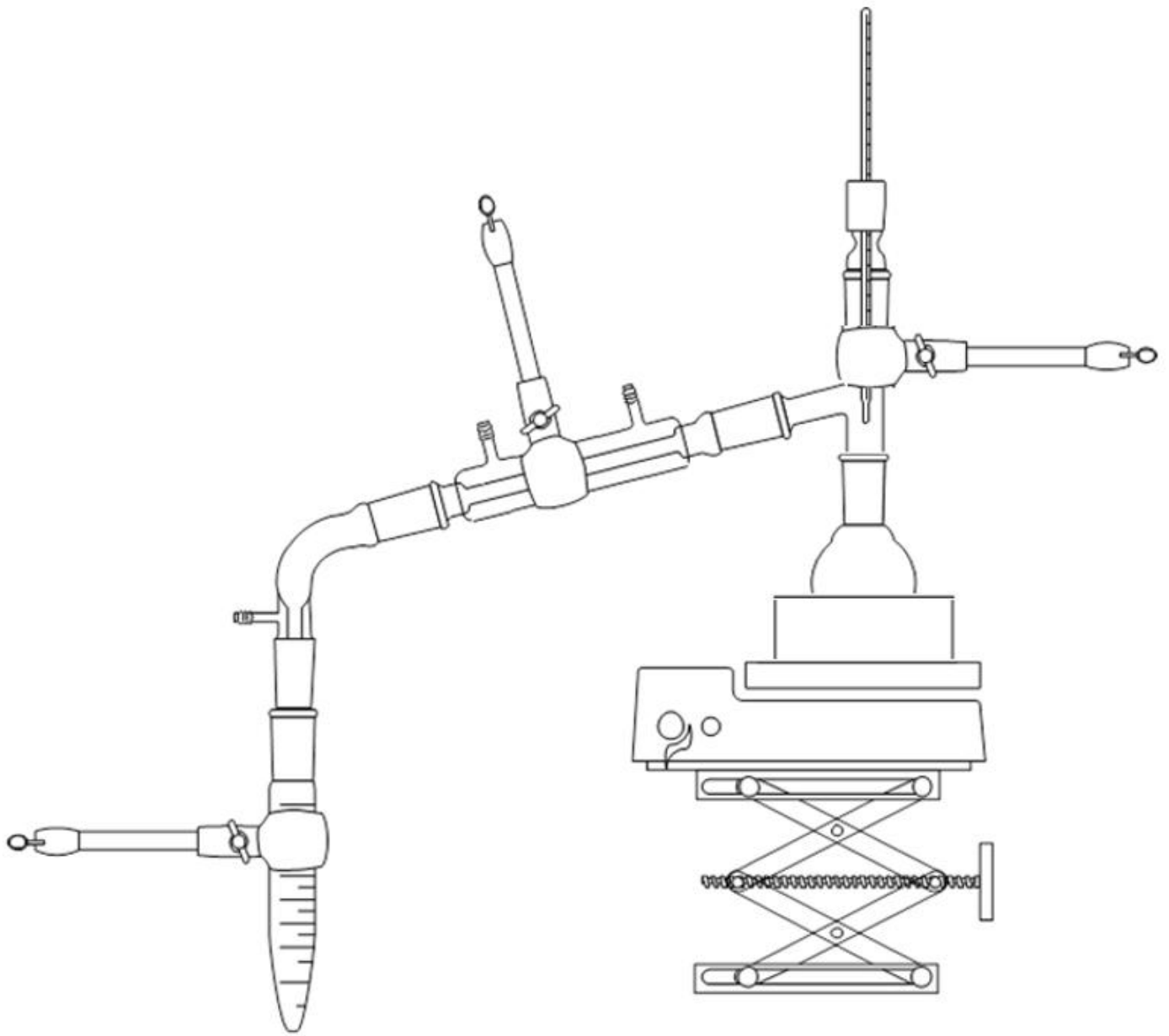
L/L 20°C



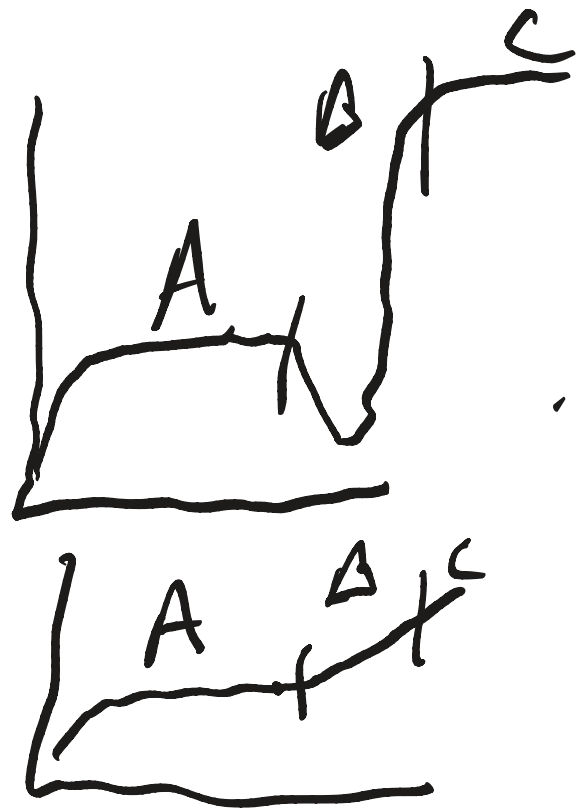
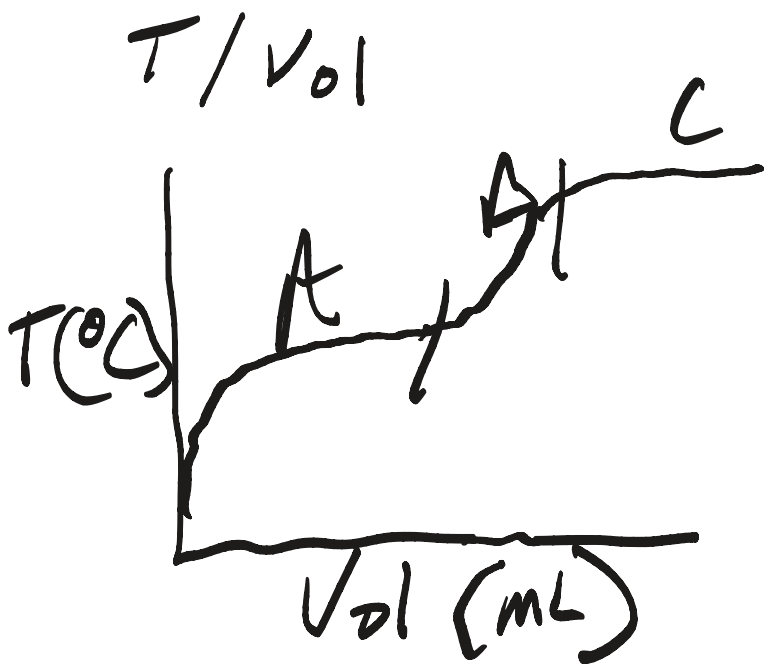
Fract.  
Distillation

30 mL acetone / toluene





Hot plate 200-230  
stir plate 4-5



Fraction	Volume (mL)	%A / %T
A	10	
B	10	
C	10	

1. acetone / Toluene identify
2. ratio?

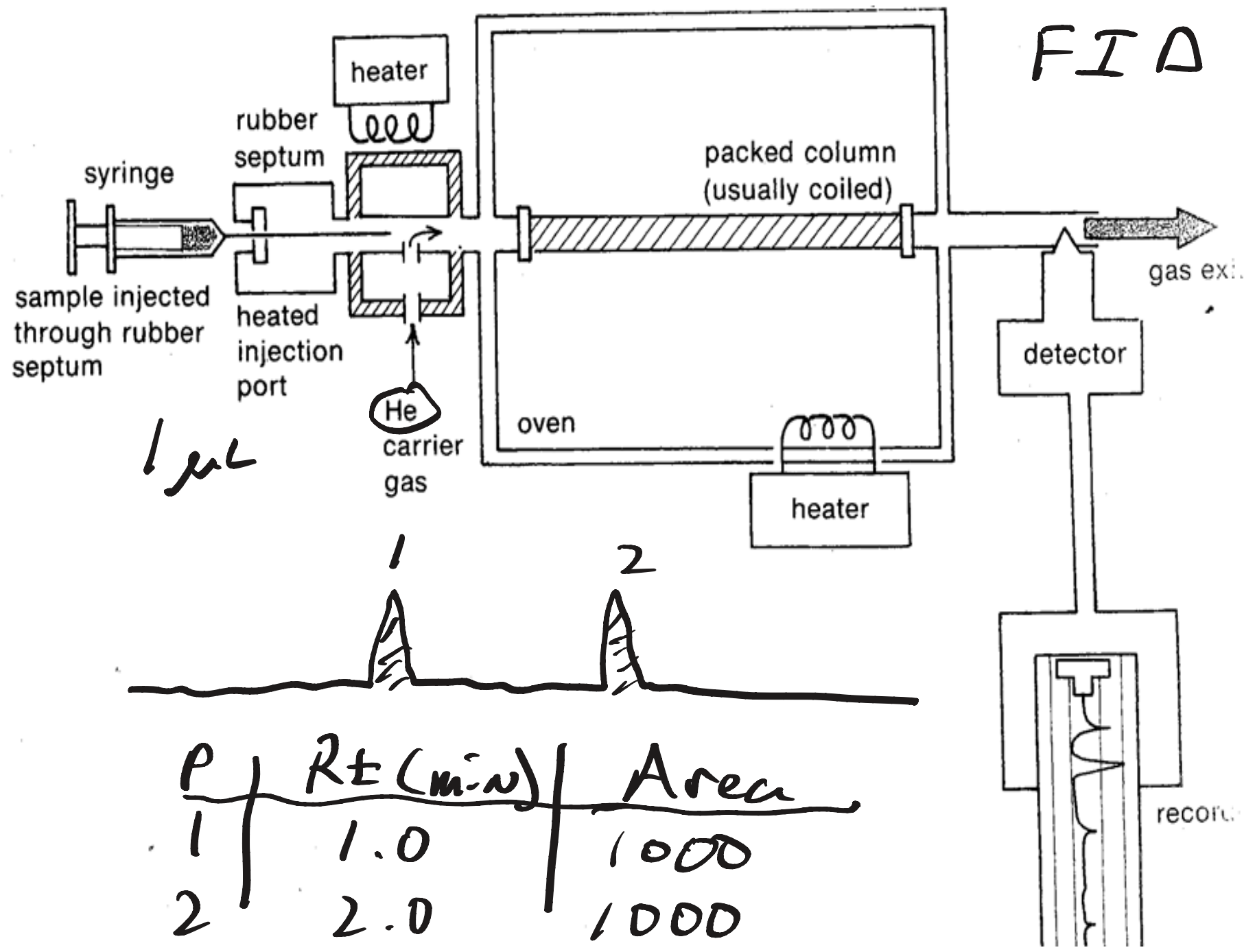
# gas chromatography

Stationary phase (A)

Mobile phase (NA)

Boiling points

FID



1.2

P	Rt (min)	Area
1	1.0	1000
2	2.0	1000

recor...

P1 1000

P2 1000

---

2000

tot. P.A.

$$\left( \frac{1000}{2000} \right) \times 100 =$$

% P.A.

F	Vd	GC %A/%T	Vol A/T
A	10	80/20	8/2
B	10	50/50	5/5
C	10	0/100	0/10
			<hr/> 13/17

$(13/30) \times 100$



